



Pergamon

Tetrahedron Letters 41 (2000) 4085–4088

TETRAHEDRON  
LETTERS

# Kinetic resolution of 1-acenaphthenol and 1-acetoxynaphthene through lipase-catalyzed acylation and hydrolysis

Louisa Aribi-Zouioueche<sup>a</sup> and Jean-Claude Fiaud<sup>b,\*</sup><sup>a</sup>Department of Organic Chemistry, University of Annaba, B.P. 12, 23000 Annaba, Algeria<sup>b</sup>Institut de Chimie Moléculaire d'Orsay, Université Paris-Sud, F-91405 Orsay, France

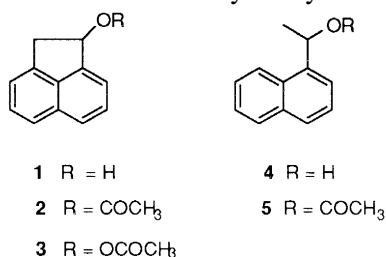
Received 7 March 2000; accepted 31 March 2000

## Abstract

Acenaphthenyl acetate and acenaphthenol are resolved through *Pseudomonas fluorescens* lipase (PFL)-catalyzed hydrolysis and acylation, respectively. By contrast, the structurally related 1-(1-naphthyl)ethyl acetate and 1-(1-naphthyl)ethanol are inactive under the same reaction conditions. © 2000 Elsevier Science Ltd. All rights reserved.

**Keywords:** kinetic resolution; enzyme; catalysis; enantioselective; acylation; hydrolysis.

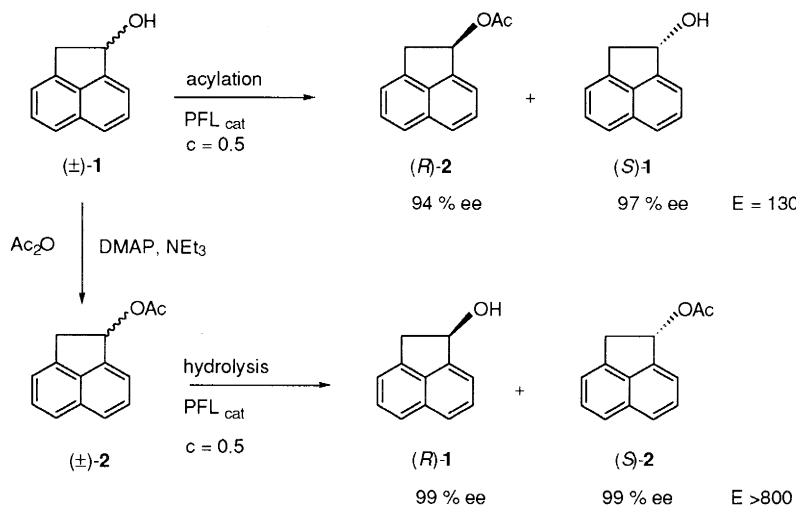
Enantiomerically pure 1-arylethanols and their derivatives are compounds of interest in transition metal-catalyzed organic synthesis. Noteworthy, 1-(1-naphthyl)ethyl acetate and carbonate have been used as chiral, optically active substrates in palladium-catalyzed substitutions.<sup>1</sup> In this context, we have been interested in the preparation of optically active 1-acenaphthenol to investigate the behaviour of its esters in Pd-catalyzed substitutions. (–)-1-Acenaphthenol **1**, a benzylic alcohol which is an early metabolite of oxidation of acenaphthene, has already been obtained by fractional crystallization of the diastereomeric camphanic esters and its configuration shown to be *R* by X-ray diffraction.<sup>2</sup>



Since the first assays for lipase-catalyzed acylation of **1** were disappointing<sup>3</sup> we turned to investigations on the resolution of (±)-**1** by enzymatic hydrolysis of the corresponding acetate and methyl carbonate (Scheme 1). Two lipases were examined: a microbial lipase from *Pseudomonas fluorescens* (PFL) and

\* Corresponding author. E-mail: fiaud@icmo.u-psud.fr (J.-C. Fiaud)

a mammalian, the rabbit gastric lipase (RGL),<sup>4</sup> since this latter showed good enantioselectivities in the kinetic resolution of secondary benzylic alcohols through acylation.<sup>3</sup> Moreover, we wished to compare both the activity and the selectivity displayed by these lipases in the kinetic resolution of **4** and **5**, which may be regarded as the flexible counterparts of rigid **1** and **2**. The results for lipase-catalyzed hydrolyses of **2**, **3** and **5** are shown in Table 1.



Scheme 1.

Table 1  
Lipase-catalyzed hydrolyses of **2**, **3** and **5**

Substrate <sup>a</sup>	Lipase (mg)	Reaction time (h)	Conversion <sup>b</sup> (%)	Recovered Ester ee(%) (% yield) <sup>c</sup>	Alcohol ee(%) <sup>d</sup> (% yield) <sup>c</sup>	E <sup>b</sup>
(±)- <b>3</b>	PFL (50)	90	41	62 <sup>e</sup> (17)	89 (13)	30
(±)- <b>3</b>	RGL (80)	120	42	8 <sup>e</sup> (22)	10 (17)	1.3
(±)- <b>2</b>	PFL (50)	90	50	99 <sup>f</sup> (31)	99 (35)	>800
(±)- <b>2</b>	RGL (100)	24	61	81 <sup>f</sup> (32)	51 (40)	8
(±)- <b>5</b>	PFL (50)	120	< 5	-	-	-
(±)- <b>5</b>	RGL (50)	120	< 5	-	-	-

<sup>a</sup> 2 mmol, 30°C. <sup>b</sup> c and E were calculated from  $ee_{\text{substrate}}$  and  $ee_{\text{product}}$  using standard equations [5]; <sup>c</sup> yields given for isolated products after silica gel chromatography (hexane / ethyl acetate 80 / 20). <sup>d</sup> measured by chiral hplc on Chiralcel<sup>®</sup> OD-H column, hexane/isopropanol 97/3, flow 0.5 mL.min<sup>-1</sup>. <sup>e</sup> measured by chiral hplc on Regis<sup>®</sup> (S,S)-WHELK-01 column, hexane/isopropanol 9/1, flow 0.5 mL.min<sup>-1</sup>. <sup>f</sup> measured by chiral hplc on Chiralcel<sup>®</sup> OD-H column, hexane/isopropanol 9/1, flow 0.5 mL.min<sup>-1</sup>.

RGL was inactive for hydrolysis of **5**, and moderately selective for hydrolysis of **2**. By contrast, although inactive for hydrolysis of **5**, PFL displayed both good activity and enantioselectivity ( $E > 800$ ) in hydrolysis of **2**.

An analogous investigation was carried out for acylation of **1** and **4**. The results are shown in Table 2.

Table 2  
Lipase-catalyzed acylation of alcohols **1** and **4** with isopropenyl acetate

Substrate <sup>a)</sup>	Lipase (mg)	Reaction time (h)	Conversion <sup>b</sup> (%)	Recovered alcohol ee(%) <sup>c</sup> (% yield) <sup>d</sup>	Ester ee(%) <sup>e</sup> (% yield) <sup>d</sup>	E <sup>b</sup>
(±)- <b>1</b>	PFL (50)	24	50	97 (34)	94 (41)	130
(±)- <b>1</b>	RGL (80)	24	90	77 (22)	8	2.2
(±)- <b>4</b>	PFL (50)	90	- <sup>f)</sup>	-	-	-
(±)- <b>4</b>	RGL (100)	120	20	19	78	11

<sup>a</sup> 2 mmol, 30°C; 3 mmol isopropenyl acetate. <sup>b</sup> c and E were calculated from ee<sub>substrate</sub> and ee<sub>product</sub> using standard equations<sup>5</sup>. <sup>c</sup> measured by chiral hplc on Chiralcel<sup>®</sup> OD-H column, hexane/isopropanol 97/3, flow 0.5 mL min<sup>-1</sup>. <sup>d</sup> yields given for isolated products after silica gel chromatography. <sup>e</sup> measured by chiral hplc on Chiralcel<sup>®</sup> OD-H column, hexane/isopropanol 9/1, flow 0.5 mL min<sup>-1</sup>. <sup>f</sup> very low conversion: less than 1 % ester produced.

Here again, both **1** and **4** were acylated with isopropenyl acetate in the presence of RGL with poor ( $E=1.8$ ) to moderate ( $E=14$ ) selectivity. The same trend as for hydrolysis was observed in PFL-catalyzed acylation: PFL was found to be inactive for acylation of **4**, but both active and enantioselective in acylation of **1**.

The chiroptical properties of **1** and **2** are reported in Table 3.

Table 3  
Chiroptical properties of **1** and **2**, obtained through acylation and hydrolysis, respectively<sup>a</sup>

Compound	$[\alpha]_D^{20}$	c (g.100mL <sup>-1</sup> )	% ee <sup>b</sup>	Compound	$[\alpha]_D^{20}$	c (g.100mL <sup>-1</sup> )	% ee <sup>b</sup>
(S)- <b>1</b>	+ 1.34	2.6	97	(R)- <b>1</b>	- 1.38	2.1	99
(R)- <b>2</b>	+ 81.6	3.4	94	(S)- <b>2</b>	- 85.2	2.2	99

<sup>a</sup> optical rotations measured with a Perkin Elmer 241 polarimeter, in CHCl<sub>3</sub> solvent.

<sup>b</sup> see Tables 1 and 2.

RGL was neither satisfactory in acylation of **1** and **4**, nor in hydrolysis of **2**, **3** and **5**. PFL was inactive in acylation of **4** and hydrolysis of **5**, but showed excellent enantiocomplementarity, since it was both active and enantioselective in acylation of **1** and hydrolysis of the corresponding esters **2** and **3**. Both enantiomers of **1** and **2** were thus obtained directly with the same enzyme.

The high  $E$  value recorded for PFL-catalyzed hydrolysis of **2** indicates that the spatial structure of the  $R$ -configured rigid substrate fits nicely in the active site of the enzyme for acylation. The same trend holds for acylation of **1**, where ( $R$ )-**1** should have the ideal shape for nucleophilic attack of the acylenzyme.

Rigid **1** and **2** can be regarded as resembling particular conformations of **4** and **5**, respectively. Since **4** and **5** are inactive, we may conclude that the active conformation of **4** and **5** for enzyme-mediated acylation and hydrolysis cannot be reached, or is too poorly populated. This sounds reasonable since in such a conformation, the methyl group of **4** and **5** would suffer severe steric interactions with the *peri*-H atom attached to the C-8 naphthalenic carbon atom. Such results could help in designing the shape of the active site of the enzyme and the steric and spatial requirements for activity and selectivity in PFL-catalyzed acylations and hydrolysis of secondary alcohols. Work is in progress along this line.

In summary, acenaphthenyl acetate and acenaphthenol are resolved and obtained in >97% *ee* through *Pseudomonas fluorescens* lipase (PFL)-catalyzed hydrolysis and acylation, respectively. (*R*)-Acenaphthol has the required structure for a nice fit with the enzyme active site. By contrast, the structurally related 1-(1-naphthyl)ethyl acetate and 1-(1-naphthyl)ethanol are inactive under the same reaction conditions, indicating that the active conformation of the (*R*)-enantiomer is of too high an energy.

## Acknowledgements

This work was financially supported by CMEP (AP-95 MDU 327). The Algerian Ministry of Education and Scientific Research is gratefully thanked.

## References

1. Legros, J.-Y.; Toffano, M.; Fiaud, J.-C. *Tetrahedron* **1995**, *51*, 3253–3246.
2. Hu, Y.; Ziffer, H.; Silverton, J. V. *Can. J. Chem.* **1989**, *67*, 60–62.
3. Legros, J.-Y.; Toffano, M.; Drayton, S. K.; Rivard, M.; Fiaud, J.-C. *Tetrahedron Lett.* **1997**, *38*, 1915–1918.
4. Moreau, H.; Gargoni, Y.; Lecat, D.; Junien, J. L.; Verger, R. *Biochem. Biophys. Acta* **1988**, *960*, 286–293.
5. Chen, C. S.; Wu, C. H.; Gidaukas, G.; Sih, C. J. *J. Am. Chem. Soc.* **1987**, *106*, 2182–2187. Kagan, H.; Fiaud, J.-C. In *Topics in Stereochemistry*; Eliel, E. L.; Wilen, S. H., Eds.; Interscience, 1988; Vol. 106, pp. 249–331.